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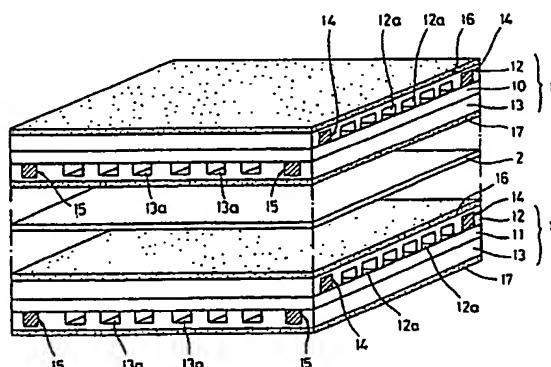
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(54) Electroconductive integrated substrate and process for producing the same.

(55) An electroconductive integrated substrate and a process for producing the same, which comprises a porous carbonaceous plate having a large number of pores and gas permeability, and a gas unpermeable electroconductive plate laminated on the porous carbonaceous plate and integrally bonded thereto by thermoplastic resin which is present substantially only in the pores of the porous carbonaceous plate. It is preferred to use thermoplastic resin having a melting viscosity of from 10^2 to 10^6 poises, and a gas permeable porous carbonaceous plate having a mean pore diameter of from 20 to 150 μm and a porosity of from 40 to 85%. The porous carbonaceous plate and the gas unpermeable electroconductive plate are laminated one upon the other with a thermoplastic resin film interposed therebetween, and the film is melted to bond the plates together. The integrated substrate obtained has an electrical resistance of not greater than 7 $\text{m}\Omega$, preferably not greater than 1.5 $\text{m}\Omega$, at the bonded surface in the

thickness direction thereof.

FIG. 1



ELECTROCONDUCTIVE INTEGRATED SUBSTRATE AND PROCESS FOR PRODUCING THE SAME

BACKGROUND OF THE INVENTION

This invention relates to an electroconductive integrated substrate and a process for producing the same which is suitable for a functional element of an electrode substrate unit, fuel or oxidative gas passage unit, cooling unit and the like for a fuel cell containing acid or alkali as an electrolyte, particularly a phosphoric acid fuel cell, or a membrane fuel cell using an ion-exchange membrane as an electrolyte, and a functional element of an electrode substrate unit etc. for secondary batteries of electrolyte circulation type.

A fuel cell is, as is known in the art, structured by stacking a plurality of electrode substrate units each comprising a gas separating plate with electroconductivity sandwiched between gas permeable electrode substrates with electroconductivity. In such unit, if the electrical contact resistance at the surface between the gas separating plate and the electrode substrate is large, the voltage drop is naturally great when a current is passed in the stacking direction and, in the case of fuel cell, the electric efficiency is low.

To overcome this problem, an electrode substrate unit has been proposed, for example, in Japanese Laid-Open Patent Publication No. 61-19069, wherein porous electrode substrates with gas permeability, made of carbon fibers, are bonded to both surfaces of a carbonaceous gas separating plate with electroconductivity by a phenolic resin and then the resulting structure is heat treated to carbonize the resin. Since the resin is carbonized, this unit has a lower electrical contact resistance as compared to a structure in which the gas separating plate and the porous electrode substrates are merely laminated. There is, however, the problem that the manufacturing cost is high because of the necessity of the heat treatment step. Moreover, since the resin shrinks during the heat treatment step, the unit may be warped or fail to provide sufficient bonding strength, causing the problem of the bonded surfaces being separated. Naturally, these inconveniences are more prominent in the case of larger-sized area unit.

In Japanese Laid-Open Patent Publication No. 62-211868 is disclosed an electroconductive gas separating plate which is fabricated by sandwiching a thermoplastic resin film between two porous carbonaceous plates with gas permeability and bonding these together with application of heat and pressure to form an integral structure. As the thermoplastic resin film, the Japanese Publication discloses using a polytetrafluoroethylene (PTFE) film.

However, because of its melting viscosity as high as 10^{10} - 10^{13} poises, PTFE, if melted with application of heat, does not penetrate satisfactorily into the pores of the gas permeable porous carbonaceous plate and thus fails to provide sufficient bonding strength. Furthermore, PTFE remains in the form of a thick layer at the bonded surfaces. Therefore, the electrical resistance at the bonded surfaces of adjacent gas permeable porous carbonaceous plates are large.

Japanese Laid-Open Patent Publication No. 63-48763 discloses an electrode substrate for a fuel cell, which is produced by bonding a porous carbonaceous plate and a separator by a "Teflon" dispersion. The "Teflon" dispersion is prepared by dispersing 10-70% by weight of polytetrafluoroethylene (PTFE), and after this dispersion is applied to the bonding surface of the separator, the porous carbonaceous plate is attached to the bonding surface of the separator and the two are bonded together with application of heat and pressure. The electrode substrate obtained in this manner is, however, poor in electroconductivity due to the use of PTFE dispersion, and it is difficult to apply PTFE uniformly to the bonding surface, thus failing to provide a desired bonding strength.

OBJECTS AND SUMMARY OF THE INVENTION

A primary object of this invention is to provide an electroconductive integrated substrate which solves the above-mentioned problems associated with conventional electroconductive integrated substrates and which has low electrical resistance at the bonded surfaces and thus is best suited for a functional element of an electrode substrate unit, gas passage unit, cooling unit and the like for various types of fuel cell.

Another object of the invention is to provide an electroconductive integrated substrate which, when used to form a unit, is free from warp and separation at the bonding surface.

Still another object of the invention is to provide a process for producing an electroconductive integrated substrate which does not require a heat treatment step after bonding, unlike the aforementioned conventional methods, and which can be manufactured at low cost.

According to the present invention, an electroconductive integrated substrate is provided which comprises a porous carbonaceous plate having a large number of gas permeable pores, and a gas unpermeable electroconductive plate laminated

on the porous carbonaceous plate and integrally bonded thereto by thermoplastic resin which is present substantially only in the pores of the porous carbonaceous plate.

Preferably, the thermoplastic resin has a melting viscosity of from 10^2 to 10^6 poises, and consists of one or more selected from the group of a copolymerizate of tetrafluoroethylene and perfluoroalkylvinylether, a copolymerizate of tetrafluoroethylene and hexafluoropropylene, polyether sulfone, polyvinyl chloride, polyvinylidene chloride, polyethylene, polypropylene, polyamide, polysulfone, polyester, polyphenylene sulfide, polyether sulfone, polyetheretherketone, polyvinylidene fluoride, fluorine-contained polyester, and fluorine-contained acrylic acid resin.

The gas permeable porous carbonaceous plate preferably has a mean pore diameter of from 20 to 150 μm and a porosity of from 40 to 85%, and comprises short carbon fibers, having a diameter of single fiber of from 4 to 15 μm and a fiber length of from 2 to 20 mm, which are randomly dispersed within substantially a 2-dimensional plane and are bonded with each other by carbon, e.g., carbonized resin.

The electroconductive integrated substrate of the invention is produced by laminating a porous carbonaceous plate having a large number of gas permeable pores and a gas unpermeable electroconductive plate one upon the other with a thermoplastic resin film; preferably having a thickness of from 10 to 200 μm , interposed therebetween, and heating the laminated structure thus obtained under pressure, thereby melting the thermoplastic resin film to bond the porous carbonaceous plate and the gas unpermeable electroconductive plate together.

The porous carbonaceous plate may integrally be bonded to only one surface of the gas unpermeable electroconductive plate or to both surfaces of the same.

The electroconductive integrated substrate according to the invention has an electrical resistance ΔR of not greater than 7 $\text{m}\Omega$, preferably not greater than 1.5 $\text{m}\Omega$, at a bonded surface thereof in the thickness direction, in accordance with the following formula:

$$\Delta R = (R - \sum_{i=1}^N R_i) / (N - 1)$$

where R and R_i represent, respectively, the electrical resistance ($\text{m}\Omega$) in the thickness direction of the bonded structure comprising j porous carbonaceous plates ($j = 1$ or more) and k gas unpermeable electroconductive plates ($k = 1$ or more),

and the electrical resistance ($\text{m}\Omega$) in the thickness direction of an i -th one of the j porous carbonaceous plates and the k gas unpermeable electroconductive plates constituting the bonded structure, and N represents the total number of the j porous carbonaceous plates and the k gas unpermeable electroconductive plates bonded together, i.e., $N = j+k$, the electrical resistance in the thickness direction being calculated from a voltage drop per unit area (1 cm^2) measured when a current of 100 mA is passed per unit area between electrodes sandwiching a plate to be measured therebetween.

The above and other objects, features and advantages of the present invention will become more apparent from the ensuing detailed description taken in conjunction with the accompanying drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

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Fig. 1 is a perspective view of an electrode substrate unit for a phosphoric acid fuel cell in which is used an electroconductive integrated substrate according to the invention;

Fig. 2 is a perspective view of a gas passage unit for a phosphoric acid fuel cell in which is used an electroconductive integrated substrate according to the invention;

Fig. 3 is a perspective view of a porous carbonaceous plate with gas permeability, used in the electrode substrate unit of and the gas passage unit of Fig. 2;

Fig. 4 is a perspective view illustrating a modification of gas permeable porous carbonaceous plate;

Fig. 5 is a perspective view illustrating how a solid rod-shaped resin is fitted in a sealing groove of the gas permeable porous carbonaceous plate of Fig. 3;

Fig. 6 is a perspective view showing a hollow rod-shaped resin to be fitted in the sealing groove of the gas permeable porous carbonaceous plate;

Fig. 7 is a front view showing part of the gas permeable porous carbonaceous plate and a resin properly fitted in the sealing groove of the plate;

Fig. 8 is a front view showing part of the gas permeable porous carbonaceous plate and a resin improperly fitted in the sealing groove;

Fig. 9 is a front view of part of the gas permeable porous carbonaceous plate, illustrating a state in which a resin fitted in the sealing groove of the plate is solidified to form a gas seal after melted;

Fig. 10 is a perspective view illustrating another embodiment of electrode substrate unit for a fuel cell in which is used an electroconductive integrated substrate according to the invention;

Fig. 11 is a perspective view illustrating a cooling unit for a fuel cell in which is used an electroconductive integrated substrate according to the invention; and

Fig. 12 is a perspective view illustrating still another embodiment of electrode substrate unit for a fuel cell in which is used an electroconductive integrated substrate according to the invention.

DETAILED DESCRIPTION

An electroconductive integrated substrate according to the present invention has a laminated structure comprising a porous carbonaceous plate with gas permeability and an electroconductive plate with gas unpermeability, the two plates being bonded together at their laminating surfaces. To bond these two plates, the gas permeable porous carbonaceous plate may be bonded to only one surface of the gas unpermeable electroconductive plate or to both surfaces of the same. On the contrary, the electroconductive plate with gas unpermeability may be bonded to one surface of the gas permeable porous carbonaceous plate. A desired laminated structure may be employed depending on the applications of the integrated substrate. Furthermore, the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate may be both in the form of a totally flat plate, or a non-flat plate obtained by mechanical machining. The shape of the plates may vary depending on their applications. The gas permeable porous carbonaceous plate is usually provided with a gas seal at opposite side portions thereof, to prevent leakage of gases therefrom.

The gas permeable porous carbonaceous plate is preferably formed by dispersing short carbon fibers (e.g., diameter of single fiber: 4-15 μm ; fiber length: 2-20 mm) in random directions within a substantially 2-dimensional plane and bonding the short fibers by carbon, e.g., carbonized resin, and has interconnected pores and thus is permeable to gases in the thickness direction as well as in the directions along the surface thereof. This type of gas permeable porous carbonaceous plate can not only achieve high strength of bonding with the gas unpermeable electroconductive plate, but also serve to lower the electrical resistance at the contact surfaces. The gas permeable porous carbonaceous plate of this type is disclosed, for example, in Japanese Patent Publications No. 53- 18603 and No. 53-43920, in which the porous carbonaceous plate is fabricated by a paper making method.

Alternatively, the gas permeable porous carbonaceous plate may be produced by molding a mixture of short carbon fibers (e.g., diameter of

single fiber: 4-15 μm ; fiber length: 0.1-1 mm) and a resin carbonizable by heat treatment step, such as phenolic resin, into a desired shape and heat treating the mixture. In this case, the short carbon fibers lie randomly in three dimensions. This type of gas permeable porous carbonaceous plate produced by a molding method is disclosed in Japanese Patent Publications No. 61-50912 and No. 62-29207, for example. The porous carbonaceous plate of this type may be used in combination with the above-mentioned porous carbonaceous plate in which short carbon fibers are dispersed 2-dimensionally, depending on the applications.

The mean pore diameter and porosity of the gas permeable porous carbonaceous plate vary depending on the diameter and quantity of the short carbon fibers used, the amount of carbonizable resin, etc. In order to ensure firm bonding with the gas unpermeable electroconductive plate and avoid the electrical resistance at the bonded surfaces being increased, the gas permeable porous carbonaceous plate preferably has a mean pore diameter ranging from 20 to 150 μm and a porosity ranging from 40 to 85%. Still preferred ranges of mean pore diameter and porosity are 20 to 60 μm and 50 to 80%, respectively.

It is noted that the term "porous carbonaceous plate" covers also an electroconductive elastic plate formed by molding a mixture of an electroconductive material, such as carbon powder, short carbon fibers or the like, and a resin into a desired shape, and a porous plate containing graphite.

The electroconductive plate with gas unpermeability, on the other hand, may be a high-density carbon plate, a glassy carbon plate, an electroconductive plastic plate formed by molding a mixture of an electroconductive material, such as carbon powder, short carbon fibers or the like, and a resin into a desired shape, a graphite plate formed by heating graphite powder which has been subjected to oxidation treatment, to obtain expanded graphite powder and then compression molding the powder, or a metal plate such as an electrolytic copper plate, an aluminum plate, or a stainless steel plate which are coated with electroconductive resin. The gas unpermeable electroconductive plate has electroconductivity but is scarcely permeable to gases in the thickness direction. The thickness of the electroconductive plate may be set suitably depending on its applications. To increase the strength of bonding with the gas permeable porous carbonaceous plate, the gas unpermeable electroconductive plate may be subjected to surface roughening treatment beforehand.

The gas permeable carbonaceous plate and the gas unpermeable electroconductive plate, described above, are bonded integrally to each other

by a thermoplastic resin consisting of one or more selected from the group of a copolymerize of tetrafluoroethylene and perfluoroalkylvinylether (PFA), a copolymerize of tetrafluoroethylene and hexafluoropropylene (PFEP), and polyether sulfone (PES), to obtain an electroconductive integrated substrate. Thus, the electroconductive integrated substrate is constructed by use of the thermoplastic resin which is thermally and chemically stable, and accordingly is most suited for constituting the aforementioned functional element of a fuel cell using an electrolyte of phosphoric acid which is corrosive at high temperatures, 200°C, for example.

In the case of a membrane fuel cell, for example, which is operated at relatively low temperature and does not require high chemical resistance, the thermoplastic resin may consist of one or a mixture of polyvinyl chloride, polyvinylidene chloride, polyethylene, polypropylene, polyamide, polysulfone, polyester, polyphenylene sulfide, polyetheretherketone, polyvinylidene fluoride, fluorine-contained polyester, and fluorine-contained acrylic acid resin.

The thermoplastic resins mentioned above have a melting viscosity of as low as 10^2 to 10^5 poises. Therefore, in the bonding step described hereinafter, the thermoplastic resin penetrates satisfactorily into the pores of the gas permeable porous carbonaceous plate and remains substantially only in the pores of the porous carbonaceous plate, thus enabling the electrical resistance at the bonded surfaces to be greatly reduced. The porous carbonaceous plate has a large number of pores and accordingly is not smooth-surfaced. Therefore, if viewed microscopically, the surface of the porous carbonaceous plate is difficult to be defined decisively, and there may be a case where the thermoplastic resin is partly left in the form of layer between the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate. Nevertheless, resin remaining in the form of layer at the bonded surfaces was not observed so clearly as in the case of using PTFE mentioned above.

The melting viscosity was measured by using a capillary viscometer under the method as defined by ASTM, D-1238. The temperature for measurement was selected from within the range of from 10 to 70°C above the melting point or softening point of the thermoplastic resin.

Using the thermoplastic resin, the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate are bonded together in the following manner.

A film made of the above-mentioned thermoplastic resin and having a thickness of from 10 to 200 μm is interposed between the gas permeable porous carbonaceous plate and the gas unper-

meable electroconductive plate, and the film is melted under pressure to bond the two plates together. The pressure applied at this time may be 5-20kgf/cm². After this step, the bonded structure is preferably cooled down to or close to room temperature and then released from the applied pressure. As the thermoplastic resin film, it is preferred to use a porous film having a plurality of small pores with a pitch of the order of several millimeters to several centimeters, since air enclosed within the bonded structure is let out through the pores during the bonding step.

In carrying out the bonding step, careful consideration must be given to the thickness of the thermoplastic resin film to be used so that the amount of the thermoplastic resin may not be unnecessarily large. If the thermoplastic resin is excessive, a thick layer of the thermoplastic resin, which is electrically insulating, is formed between the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate, thus causing an increase of the electrical resistance at the bonded surfaces.

Preferably, to improve the bonding strength between the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate and also to avoid the electrical resistance at the bonded surfaces being increased, the gas permeable porous carbonaceous plate have a mean pore diameter ranging from 20 to 150 μm and a porosity ranging from 40 to 80%, as mentioned above.

If the mean pore diameter is less than 20 μm , the molten thermoplastic resin may not penetrate sufficiently into the gas permeable porous carbonaceous plate; whereas if the mean pore diameter exceeds 150 μm , recesses may be formed in the thermoplastic resin corresponding in position to the pores of the gas permeable porous carbonaceous plate and thus the bonding strength obtained will be insufficient, though this phenomenon depends on the amount of the thermoplastic resin used. The pore diameter can be measured with a mercury porosimeter, conventionally known. The mean pore diameter is here defined as 50% of the accumulated pore volume.

The electroconductive integrated substrate of the present invention has electrical resistance ΔR of not greater than 7 $\text{m}\Omega$, preferably not greater than 1.5 $\text{m}\Omega$, at the bonded surfaces in the thickness direction. The electrical resistance in the thickness direction was calculated from a voltage drop per unit area (1 cm²) measured when a current of 100 mA was passed per unit area between mercury electrodes sandwiching an object to be measured therebetween. This procedure was followed in measurement of the electrical resistance in the thickness direction of the gas permeable

porous carbonaceous plate and the gas unpermeable electroconductive plate, described hereinafter, unless otherwise explicitly defined. The electrical resistance ΔR of the electroconductive integrated substrate at the bonding surface thereof is defined as

$$\Delta R = (R - \sum_{i=1}^N R_i) / (N - 1)$$

where R and R_i represent, respectively, the electrical resistance ($m\Omega$) in the thickness direction of the bonded structure comprising j porous carbonaceous plates ($j = 1$ or more) and k gas unpermeable electroconductive plates ($k = 1$ or more), and the electrical resistance ($m\Omega$) in the thickness direction of an i -th one of the j porous carbonaceous plates and the k gas unpermeable electroconductive plates constituting the bonded structure, and N represents the total number of the j porous carbonaceous plates and the k gas unpermeable electroconductive plates bonded together, i.e., $N = j+k$.

In order to improve the bonding strength, the aforementioned porous carbonaceous plate in which short carbon fibers are dispersed randomly substantially within a 2-dimensional plane is preferably used. If such porous carbonaceous plate is used, the thermoplastic resin, when introduced into the pores, surrounds the carbon fibers and bonds the porous carbonaceous plate to the gas unpermeable electroconductive plate with a bridge-like structure. Thus, not only firm bonding is ensured even with a small amount of thermoplastic resin but the electrical resistance at the bonded surfaces can be drastically lowered.

Preferred embodiments of the present invention will now be described.

Fig. 1 shows electrode substrate units 1 for a phosphoric acid fuel cell, each unit comprising a planar gas unpermeable electroconductive plate 10 and gas permeable porous carbonaceous plates 12 and 13 bonded to respective opposite surfaces of the electroconductive plate 10. The carbonaceous plates 12 and 13 have a plurality of parallel grooves 12a and 13a for gas passage, respectively, formed in one surface thereof bonded to the corresponding surface of the electroconductive plate 10, and have the other surface thereof (the surface contacting a matrix layer 2 referred to later) provided with catalyst layers 16 and 17, respectively, containing a platinum or platinum series catalyst or the like. The carbonaceous plates 12 and 13 are bonded to the plate 10 such that the grooves 12a of the plate 12 are at right angles to the grooves 13a of the plate 13. The phosphoric acid fuel cell is

constructed by stacking a multiplicity of such electrode substrate units 1 provided with catalyst layers, with a matrix layer 2, which contains a phosphoric acid electrolyte, interposed between the adjacent units. One of the two carbonaceous plates, i.e., the plate 12, serves as a gas permeable electrode substrate for an oxidative gas (usually, air), and the other carbonaceous plate 13 serves as a gas permeable electrode substrate for a fuel gas (hydrogen gas etc.); whereas the electroconductive plate 10 serves as a separator for separating the two gases, i.e., the oxidative gas and the fuel gas. The oxidative gas and the fuel gas are introduced respectively into the plural grooves 12a and 13a of the carbonaceous plates 12 and 13. The carbonaceous plates 12 and 13 are respectively provided with gas seals 14 and 15 located at opposite side portions thereof and extending parallel to their respective grooves 12a and 13a.

In the above embodiment, like the embodiments to be described hereinafter, the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate are bonded together by the aforementioned thermoplastic resin which is present substantially solely in the pores of the gas permeable porous carbonaceous plate, or by such resin in the pores plus the thermoplastic resin in the form of very thin layer present between the two plates. The thermoplastic resin layer, however, cannot be observed so clearly as in the case of using PTFE, as mentioned above.

Fig. 2 shows gas passage units 3 for a phosphoric acid fuel cell, each consists of a planar gas unpermeable electroconductive plate 30 and gas permeable porous carbonaceous plates 32 and 33 respectively bonded to each of opposite surfaces of the electroconductive plate 30. The carbonaceous plates 32 and 33 have plural parallel gas passage grooves 32a and 33a, respectively, formed in one surface thereof opposite to that bonded to the electroconductive plate 30, and are bonded to the plate 30 in such a manner that the grooves 32a of the carbonaceous plate 32 are at right angles to the grooves 33a of the plate 33. The phosphoric acid fuel cell using the gas passage unit 3 is constructed by repeatedly stacking the gas passage unit 3, an air electrode 5, a matrix layer 4, and a fuel electrode 6 in the order mentioned. The air electrode 5 and the fuel electrode 6 include electrode substrates 5a and 6a, respectively, carrying catalyst layers 5b and 6b on the surface thereof facing the matrix layer 4. An oxidative gas and a fuel gas are introduced respectively in the plural grooves 33a and 32a of the carbonaceous plates 33 and 32.

The carbonaceous plates 32 and 33 are provided with gas seals 34 and 35, respectively, located at opposite side portions thereof and extend-

ing parallel to their respective grooves 32a and 33a.

The gas seals arranged at the opposing side portions of the gas permeable porous carbonaceous plate will be described in detail. The gas seals are important for preventing leakage of gases, particularly the fuel gas, from the opposite sides of the gas permeable porous carbonaceous plate. Fig. 3 shows the gas permeable porous carbonaceous plate 32 used in the electrode substrate unit 1 shown in Fig. 1 and the gas passage unit 3 shown in Fig. 2. As illustrated, two grooves 32b are formed in opposite side portions of the carbonaceous plate 32 such that they extend parallel to opposing two sides as well as the grooves 32a of the plate 32. If the width of the groove 32b is too broad, the effective area of the carbonaceous plate working as the electrode is reduced, and therefore, a larger amount of resin is required for filling the groove for sealing, thus causing inconveniences such as low cell performance and high cost. On the other hand, too narrow groove width incurs difficulty in filling the resin. Taking these into account, the sealing grooves 32b preferably have a width of from 0.5 to 15 mm, preferably from 1 to 10 mm; a still preferred width is in the range of from 1 to 3 mm.

Depth d1 of the groove 32b may be varied depending on the thickness of the carbonaceous plate 32. To ensure the strength of the carbonaceous plate 32 and facilitate its handling, a web portion 32c appearing in the figure should have a thickness d2 of not smaller than 0.1 mm, preferably, not smaller than 0.2 mm. To increase the sealability of the seals 34, it is essential for the web portion 32c to be sufficiently impregnated with the thermoplastic resin used for bonding the carbonaceous plate 32 and the gas unpermeable electroconductive plate 30 together. In view of this, the thickness d2 of the web portion 32c should be as small as not greater than 1 mm, preferably not greater than 0.7 mm. As the sealing groove, a single groove may be formed in each of opposite side portions of the carbonaceous plate 32, as shown in Fig. 3; alternatively, two or more grooves for gas sealing (32b, 32b') may be formed in each of the opposite side portions, as illustrated in Fig. 4. The sealing grooves 32b can be cut together with the grooves 32a used for gas passage.

Then, rod-shaped thermoplastic resins to be fitted in the grooves 32b of the carbonaceous plate 32 are prepared. The same resin as used for bonding the gas permeable porous carbonaceous plate and the gas unpermeable electroconductive plate, mentioned above, can be used for this purpose. The cross-sectional shape of the rod-like resin filler is not limited in particular; for example, the resin may be formed into a solid body 36, as

shown in Fig. 5, or into a hollow body 37, as shown in Fig. 6. Although not shown, the resin may have a U- or C-shaped cross section. In any case, the resin rod should have such an outer diameter (d3, d3') that it can be received in the sealing groove 32b having the dimensions defined above

The amount of the resin to be filled must be determined taking account of the amount of resin solidified in the sealing groove 32b, as well as the amount of resin penetrated into the carbonaceous plate 32 and solidified therein. In the case of using the solid resin filler 36, its cross-sectional area may preferably be 1.01-1.4 times the cross-sectional area of the sealing groove 32b; whereas in the case of the hollow resin filler 37 shown in Fig. 6, the cross-sectional area thereof may preferably be 0.8-1.2 times that of the sealing groove 32b, because the hollow portion 37a can be left even after the sealing step.

Subsequently, the solid resin 36 or hollow resin 37, prepared in the above manner, is fitted in each sealing groove 32b. In this case, the filler 37 may partly project upward from the surface of the carbonaceous plate 32, as shown in Fig. 7; but the situation must be avoided where the filler 37, as shown in Fig. 8, protrudes widthwise beyond the width of the sealing groove 32b. If the filler 37 protrudes in this manner, the surface of the carbonaceous plate 32 will be swollen excessively or too much resin will be penetrated into the surface region of the carbonaceous plate during the bonding step. Thus, if the resin contracts while it is solidified, the integrated substrate may be warped, for example, which leads to degradation of sealability.

Then, the resin filled in this manner is heated under pressure so that the carbonaceous plate 32 is impregnated with part of the molten resin. This melt impregnation step is carried out under pressure using a press heated to a temperature at which the resin is melted, or to a temperature close thereto. The pressure applied by the press at this time may be 1-50 kgf/cm². The sealing portion 34 obtained in this manner is illustrated in Fig. 9. The molten resin is solidified in the sealing groove 32b and also in the web portion 32c impregnated with the resin; and the solidified resin, in cooperation with the resin penetrated into the web portion 32c from the bonded surfaces 30a of the carbonaceous plate 32 and the electroconductive plate 30 during the bonding step and solidified there, serves to prevent leakage of gases from the sides of the carbonaceous plate 32.

Since the resin filled in the sealing groove 32b has a shape of hollow or solid rod, the sealing groove 32b can be filled with a constant amount of resin without non-uniformity. Further, the resin can be easily fitted in the sealing groove 32b, and thus

other portions than the grooves are not contaminated with the resin and handling of the resin is facilitated. To avoid the situation that because of excessive resin filled in the groove, the sealing groove is destroyed by the resin while the resin is heated under pressure, the hollow resin rod 37 shown in Fig. 6 can advantageously be used since the cavity 37a thereof serves to absorb variation of volume in some degree and facilitate sealing. The bonding step wherein the carbonaceous plate 32 and the electroconductive plate 30 are bonded together and the melt impregnation step wherein the carbonaceous plate 32 is impregnated with the resin filled in the groove 32b can be carried out at the same time; in this case, the fabrication process can be simplified and the costs can be reduced.

Fig. 10 illustrates an electrode substrate unit 7 for a fuel cell, comprising a gas unpermeable electroconductive plate 70 having plural parallel grooves 70a and 70b for gas passage on opposite surfaces thereof, and planar gas permeable porous carbonaceous plates 72 and 73 bonded to the respective surfaces of the electroconductive plate 70. The grooves 70a formed in one surface of the gas unpermeable electroconductive plate 70 are at right angles to the grooves 70b formed in the other surface thereof. In the electrode substrate unit 7 of this embodiment, the electroconductive plate 70 functions as a separator for separating the fuel gas from the oxidative gas, and one of the carbonaceous plates, i.e., the plate 72, serves as a gas permeable electrode substrate for the fuel gas, whereas the other carbonaceous plate 73 serves as a gas permeable electrode substrate for the oxidative gas.

Fig. 11 shows a cooling unit 8 for a fuel cell, which comprises a planar gas unpermeable electroconductive plate 80, and a gas permeable porous carbonaceous plate 81 bonded at one surface thereof to the electroconductive plate 80 and having numerous U-shaped parallel grooves 81a formed in the surface thereof opposite to the bonded surface. In use, a tube, not shown, for passing a cooling medium therethrough is placed in each groove 81a. In assembling a fuel cell, the cooling unit 8 is interposed between adjacent stacks of cell units, each stack including several cell units, to provide the function of removing heat generated on electrochemical reaction to outside.

Fig. 12 shows an electrode substrate unit 9 for a fuel cell, using a combination of the aforementioned gas permeable porous carbonaceous plate 91 in which short carbon fibers are randomly dispersed within a 2-dimensional plane, and a gas permeable porous carbonaceous plates 92 and 93 in which short carbon fibers are dispersed randomly in three dimensions. The porous carbonaceous plate 91, in which short carbon fibers are

randomly dispersed within a 2-dimensional plane, is bonded to opposite surfaces of the gas unpermeable electroconductive plate 90, and the porous carbonaceous plates 92 and 93 in which short carbon fibers are dispersed randomly in three dimensions are bonded to the respective carbonaceous plates 91. The carbonaceous plates 92 and 93 have plural parallel grooves 92a and 93a for gas passage, respectively, formed in the surface thereof bonded to the plate 91, and are disposed such that the grooves 92 are at right angles to the grooves 93a. In the same manner as described with reference to Fig. 1, the carbonaceous plates 91 are bonded to the electroconductive plate 90 and the carbonaceous plates 92 and 93 are bonded to the respective carbonaceous plates 91.

EXAMPLES

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EXAMPLE 1

25 Polyacrylonitrile carbon fiber "TORAYCA" T300 (mean diameter of single filament: 7 μm ; number of filaments: 6000), manufactured by Toray Industries, Inc., was cut to 12 mm long, and the cut fibers were mixed with a paper making medium containing polyvinyl alcohol as a paper making binder, stirred, and then subjected to paper making to obtain an intermediate.

30 Subsequently, the intermediate was impregnated with 10 parts by weight of methanolic solution of phenolic resin, to obtain attachment of 150 parts by weight of phenolic resin based on 100 parts by weight of the intermediate. The prepreg was dried at 90 $^{\circ}\text{C}$, and then is heated at 170 $^{\circ}\text{C}$ under pressure of 5 kgf/cm² for 15 minutes to cure the phenolic resin.

35 Next, the prepreg with the cured phenolic resin was heat treated at 2500 $^{\circ}\text{C}$ in argon gas atmosphere to carbonize the phenolic resin, thereby obtaining a gas permeable porous carbonaceous plate with a thickness of 2 mm. This carbonaceous plate had a mean pore diameter of about 40 μm , a porosity of about 72%, and an electrical resistance of 11.1 m Ω in the thickness direction.

40 Then, between two gas permeable porous carbonaceous plates obtained in the above manner, a glassy carbon plate GCR-101 as the gas unpermeable electroconductive plate (thickness: 1.5 mm; electrical resistance in the thickness direction: 1.22 m Ω), manufactured by Kobe Steel, Ltd., was placed with a PFA film "TOYOFLOW" (thickness: 50 μm), manufactured by Toray Gōsei Film K.K., interposed between the glassy carbon plate and the respective porous carbonaceous plate. The lami-

nated structure was heated at 310°C under pressure of 10 kgf/cm² to integrally bond the gas permeable porous carbonaceous plates and the glassy carbon plate together, thereby obtaining a gas passage unit for a fuel cell as shown in Fig. 2. In this example, no grooves were formed in the unit.

The electrical resistance of the obtained gas passage unit in the thickness direction, measured in accordance with the aforementioned measurement procedure, was 23.5 mΩ which is almost equal to the total resistance of the gas permeable porous carbonaceous plates and the glassy carbon plate, i.e., 23.4 mΩ. The gas passage unit was evaluated as to the impregnation capability of phosphoric acid electrolyte, and it was found that the unit could be impregnated with 0.248 g/cm² of phosphoric acid and therefore was most suited as the gas passage unit with a function of electrolyte reservoir for a phosphoric acid fuel cell.

EXAMPLE 2

A plurality of parallel grooves with a width of 2 mm, a depth of 1 mm, and with a pitch of 2 mm between adjacent ones were formed in one surface of the gas permeable porous carbonaceous plate obtained in EXAMPLE 1. The electrical resistance of the gas permeable porous carbonaceous plate in the thickness direction was 16.6 mΩ. The electrical resistance was measured under the condition that the gas permeable porous carbonaceous plate was sandwiched between carbonaceous electrodes and a current of 100 mA per unit area (1 cm²) was passed between the two electrodes under pressure of 5 kgf/cm², and was obtained by subtracting a voltage drop produced at the contact surfaces of the gas permeable porous carbonaceous plate and the carbonaceous electrodes, from a voltage drop per 1 cm². Then, following the same procedures as those in EXAMPLE 1, an electrode substrate unit for a fuel cell as shown in Fig. 1 was obtained.

The obtained electrode substrate unit has an electrical resistance of 34.6 mΩ in the thickness direction, which is almost equal to the total resistance 34.4 mΩ of the gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 3

Following the same steps as in EXAMPLE 1, two gas permeable porous carbonaceous plates with a thickness of 0.4 mm and an electrical resistance of 2.20 mΩ in the thickness direction were prepared. As the gas unpermeable electroconductive plate, a graphite plate T2172 (thickness: 5 mm), manufactured by Tokai Carbon Co., Ltd., was

used, and grooves were formed in both surfaces of the graphite plate, as in EXAMPLE 2. The grooved graphite plate had an electrical resistance of 0.80 mΩ in the thickness direction.

Next, the gas permeable porous carbonaceous plates and the grooved graphite plate were bonded together, as in EXAMPLE 1, thereby obtaining an electrode substrate unit for a fuel cell as shown in Fig. 10.

The electrode substrate unit obtained had an electrical resistance of 5.50 mΩ in the thickness direction, which almost equals the total resistance 5.20 mΩ of the gas permeable porous carbonaceous plates and the graphite plate.

15

EXAMPLE 4

A pitch carbon fiber (mean diameter of single filament: 12 μm) manufactured by Union Carbide Corp., U.S.A., was cut to 6 mm long; and, following the same procedures as those in EXAMPLE 1, a gas permeable porous carbonaceous plate with a thickness of 2 mm was prepared. The carbonaceous plate had a mean pore diameter of about 90 μm, a porosity of about 75%, and an electrical resistance of 13.8 mΩ in the thickness direction. Subsequently, using the gas permeable porous carbonaceous plates thus obtained, an electrode substrate unit with no grooves for a fuel cell was obtained in the same manner as in EXAMPLE 1.

The electrode substrate unit obtained had an electrical resistance of 30.9 mΩ in the thickness direction almost equal to the total resistance 28.8 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 5

The carbon fiber used in EXAMPLE 4 was milled to a small length which is not greater than 1 mm, and the milled carbon fibers were dispersed in a solution of 45 parts by weight of phenolic resin based on 100 parts by weight of methanol, mixed, filtered, and then dried at 70°C.

Next, the dried mixture was pulverized, and 60 parts by weight of granular sugar that had been sifted to be 150-250 mesh were added based on 100 parts by weight of the mixture, followed by mixing. The mixture was filled in a mold and subjected to molding at 140°C under pressure of 70 kgf/cm², and the molded structure was left to stand in warm water of 80°C for 4 hours, and then is heat treated at 2500°C, thereby obtaining a gas permeable porous carbonaceous plate.

The obtained gas permeable porous carbonaceous plate in which short carbon fibers are ran-

domly scattered in three dimensions had a thickness of 2.2 mm, a mean pore diameter of 50 μm , and a porosity of 63%.

Subsequently, plural parallel grooves with a width of 2 mm, a depth of 1.1 mm and with a pitch of 2 mm between adjacent ones were formed in one surface of the above gas permeable porous carbonaceous plate. The grooved carbonaceous plate had an electrical resistance of 18.1 $\text{m}\Omega$ in the thickness direction.

Then, following the same procedures as those in EXAMPLE 1, an electrode substrate unit for a fuel cell as shown in Fig. 1 was obtained.

The electrical resistance of the obtained electrode substrate unit in the thickness direction was 40.7 $\text{m}\Omega$. The total resistance of the two gas permeable porous carbonaceous plates and the glassy carbon plate was 37.4 $\text{m}\Omega$.

EXAMPLE 6

Using a gas permeable porous carbonaceous plate with a thickness of 0.1 mm (electrical resistance in the thickness direction: 0.55 $\text{m}\Omega$), obtained in the same manner as in EXAMPLE 1, and the gas permeable porous carbonaceous plate used in EXAMPLE 5, an electrode substrate unit for a fuel cell as shown in Fig. 12 was obtained, following the same procedures as those in EXAMPLE 1.

The electrode substrate unit obtained had an electrical resistance of 42.0 $\text{m}\Omega$ in the thickness direction, which almost equals the total resistance 38.5 $\text{m}\Omega$ of the gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 7

Similar to EXAMPLE 1, a gas permeable electroconductive plate with a thickness of 3 mm and an electrical resistance of 21.9 $\text{m}\Omega$ in the thickness direction was obtained. The electroconductive plate thus obtained was laminated on a glassy carbon plate GCR-101 (thickness: 1.5 mm; electrical resistance in the thickness direction: 1.22 $\text{m}\Omega$), manufactured by Kobe Steel, Ltd., with a PFA film "TOYOFLOLON" 50P (thickness: 100 μm), manufactured by Toray Gōsei Film K.K., interposed therebetween. Then, the same procedures as those in EXAMPLE 1 were followed to obtain a cooling unit as shown in Fig. 11.

The obtained cooling unit had an electrical resistance of 23.4 $\text{m}\Omega$ in the thickness direction which almost equals the total resistance 23.1 $\text{m}\Omega$ of the gas permeable porous carbonaceous plate and the glassy carbon.

EXAMPLE 8

Following the same procedures as those in EXAMPLE 1 except that a PFEP film "TOYOFLOLON" 50F (thickness: 50 μm), manufactured by Toray Gōsei Film K.K., was used instead of PFA film, and that the bonding step was conducted at 320 °C under pressure of 10 kgf/cm², a gas passage unit as shown in Fig. 2 was obtained. No grooves were formed in the unit.

The obtained gas passage unit had an electrical resistance of 24.6 $\text{m}\Omega$ in the thickness direction almost equal to the total resistance 23.4 $\text{m}\Omega$ of the gas permeable porous carbonaceous plate and the glassy carbon plate.

EXAMPLE 9

20 A PES film "TALPA" 1000 (thickness: 50 μm), manufactured by MITSUI TOATSU CHEMICALS, INC., was used instead of PFA film, and the bonding step was conducted at 350 °C under pressure of 10 kgf/cm². Except for these, the same procedures as those in EXAMPLE 1 were followed to obtain a gas passage unit for a fuel cell as shown in Fig. 2. No grooves were formed.

25 The obtained gas passage unit had an electrical resistance of 24.1 $\text{m}\Omega$ in the thickness direction which is almost equal to the total resistance 23.4 $\text{m}\Omega$ of the gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 10

30 The same procedures as in EXAMPLE 1 were followed except for the use of a PFA film with a thickness of 25 μm and for the bonding step which was conducted at 350 °C under pressure of 10 kgf/cm², to obtain a gas passage unit for a fuel cell as shown in Fig. 2. In this case, no grooves were formed.

35 The obtained gas passage unit had an electrical resistance of 23.4 $\text{m}\Omega$ in the thickness direction almost equal to the total resistance 23.4 $\text{m}\Omega$ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 11

40 Instead of using a PFA film, a polypropylene film "TORAYFAN" #3300 (thickness: 50 μm), manufactured by Toray Gōsei Film K.K., was used, and, following the same steps as in EXAMPLE 1 except that the bonding step was conducted at 180 °C under pressure of 15 kgf/cm², a gas pas-

sage unit for a fuel cell as shown in Fig. 2 was obtained. No grooves were formed.

The obtained gas passage unit had an electrical resistance of 24.7 mΩ in the thickness direction which almost equals the total resistance 23.4 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 12

A polyethylene film (thickness: 50 μm) manufactured by Sankyo Polyethylene K.K. was used instead of PFA film, and, following the same procedures as those in EXAMPLE 1 except that the bonding step was conducted at 200°C under pressure of 15 kgf/cm², a gas passage unit for a fuel cell as shown in Fig. 2 was obtained. No grooves were formed in the unit.

The obtained gas passage unit had an electrical resistance of 23.6 mΩ in the thickness direction almost equal to the total resistance 23.4 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 13

Instead of using a PFA film, a polyvinylidene fluoride film KF SHEET P20 (thickness: 50 μm), manufactured by Kureha Chemical Industry Co., Ltd., was used, and the bonding step was conducted at 250°C under pressure of 10 kgf/cm². Except for these, the same procedures as those in EXAMPLE 1 were followed to obtain a gas passage unit for a fuel cell as shown in Fig. 2. No grooves were formed.

The gas passage unit obtained had an electrical resistance of 23.9 mΩ in the thickness direction almost equal to the total resistance 23.4 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

EXAMPLE 14

An electrolytic copper foil DT-70 as the gas unpermeable electroconductive plate (thickness: 70 μm; electrical resistance in the thickness direction: 0.01 mΩ), manufactured by MITSUI MINING AND SMELTING CO., LTD., was sandwiched between two gas permeable porous carbonaceous plates used in EXAMPLE 1, with a polypropylene film "TORAYFAN" #3300 (thickness: 25 μm), manufactured by Toray Gōsei Film K.K., interposed between the foil and each carbonaceous plate. Except that the bonding step was conducted at 180°C under pressure of 15 kgf/cm², the procedures simi-

lar to those of EXAMPLE 1 were followed, to obtain a gas passage unit for a fuel cell, as shown in Fig. 2, with no grooves formed therein.

The obtained gas passage unit had an electrical resistance of 23.8 mΩ in the thickness direction, which nearly equals the total resistance 22.2 mΩ of the two gas permeable porous carbonaceous plates and the electrolytic copper foil.

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EXAMPLE 15

A polyamide film "RAYFAN" T1401#50 (thickness: 50 μm), manufactured by Toray Gōsei Film K.K., was used instead of PFA film, and, following procedures similar to those of EXAMPLE 1 except for the bonding step which was conducted at 250°C under pressure of 10 kgf/cm², a gas passage unit for a fuel cell as shown in Fig. 2 was obtained. No grooves were formed in the unit.

The gas passage unit obtained had an electrical resistance of 24.8 mΩ in the thickness direction which almost equals the total resistance 23.4 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

25

EXAMPLE 16

Following the same procedures as those in EXAMPLE 1 except that a polyester film "LUMIRROR" #25T70 (thickness: 25 μm), manufactured by Toray Industries, Inc., was used instead of PFA film and that the bonding step was conducted at 250°C under pressure of 10 kgf/cm², a gas passage unit for a fuel cell as shown in Fig. 2 was obtained, with no grooves formed therein.

The obtained gas passage unit had an electrical resistance of 24.5 mΩ in the thickness direction which nearly equals the total resistance 23.4 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

45

EXAMPLE 17

A polyetheretherketone film "TALPA" 2000 (thickness: 50 μm), manufactured by MITSUI TOATSU CHEMICALS, INC., was used instead of PFA film, and the bonding step was conducted at 250°C under pressure of 10 kgf/cm². Except for these, the same procedures as those in EXAMPLE 1 were followed to obtain a gas passage unit for a fuel cell as shown in Fig. 2. No grooves were formed.

The gas passage unit obtained had an electrical resistance of 24.6 mΩ in the thickness direction which is almost equal to the total resistance

23.4 mΩ of the two gas permeable porous carbonaceous plates and the glassy carbon plate.

Claims

1. An electroconductive integrated substrate comprising:

a porous carbonaceous plate having a large number of pores and gas permeability; and a gas unpermeable electroconductive plate laminated on said porous carbonaceous plate and integrally bonded thereto by thermoplastic resin which is present substantially only in the pores of said porous carbonaceous plate.

2. The electroconductive integrated substrate according to claim 1, wherein said thermoplastic resin consists of one or more selected from the group of a copolymerizate of tetrafluoroethylene and perfluoroalkylvinylether, a copolymerizate of tetrafluoroethylene and hexafluoropropylene, polyether sulfone, polyvinyl chloride, polyvinylidene chloride, polyethylene, polypropylene, polyamide, polysulfone, polyester, polyphenylene sulfide, polyether sulfone, polyetheretherketone, polyvinylidene fluoride, fluorine-contained polyester, and fluorine-contained acrylic acid resin.

3. The electroconductive integrated substrate according to claim 1, wherein said porous carbonaceous plate has a mean pore diameter of from 20 to 150 μm and a porosity of from 40 to 85%.

4. The electroconductive integrated substrate according to claim 3, wherein said porous carbonaceous plate comprises short carbon fibers, having a diameter of single fiber of from 4 to 15 μm and a fiber length of from 2 to 20 mm, which are randomly dispersed within substantially a 2-dimensional plane and are bonded with each other by carbonized-resin.

5. The electroconductive integrated substrate according to claim 1, which has an electrical resistance ΔR of not greater than 7 mΩ at bonded surface thereof in the thickness direction, in accordance with the following formula:

$$\Delta R = (R - \sum_{i=1}^N R_i) / (N - 1)$$

where R and R_i represent, respectively, the electrical resistance (mΩ) in the thickness direction of the bonded structure comprising j porous carbonaceous plates (j = 1 or more) and k gas unpermeable electroconductive plates (k = 1 or more), and the electrical resistance (mΩ) in the thickness direction of an i-th one of the j porous carbonaceous plates and the k gas unpermeable elec-

troconductive plates constituting said bonded structure, and N represents the total number of the j porous carbonaceous plates and the k gas unpermeable electroconductive plates bonded together, i.e., N = j+k, said electrical resistance in the thickness direction being calculated from a voltage drop per unit area (1 cm²) measured when a current of 100 mA is passed per unit area between electrodes sandwiching a plate to be measured therebetween.

6. The electroconductive integrated substrate according to claim 1, which comprises gas seal means arranged in parallel with each of at least two opposite side edges of said porous carbonaceous plate, for preventing leakage of gases.

7. The electroconductive integrated substrate according to claim 1, which is used as a functional element for a fuel cell.

8. A process for producing an electroconductive integrated substrate, comprising the steps of: preparing a porous carbonaceous plate having a large number of pores and gas permeability; preparing a gas unpermeable electroconductive plate;

25 laminating said porous carbonaceous plate and said gas unpermeable electroconductive plate one upon the other with a thermoplastic resin film interposed therebetween; and heating the laminated structure thus obtained under pressure, thereby melting said thermoplastic resin film to bond said porous carbonaceous plate and said gas unpermeable electroconductive plate together.

35 9. The process according to claim 8, wherein said thermoplastic resin film consists of one or more selected from the group of a copolymerizate of tetrafluoroethylene and perfluoroalkylvinylether, a copolymerizate of tetrafluoroethylene and hexafluoropropylene, polyether sulfone, polyvinyl chloride, polyvinylidene chloride, polyethylene, polypropylene, polyamide, polysulfone, polyester, polyphenylene sulfide, polyether sulfone, polyetheretherketone, polyvinylidene fluoride, fluorine-contained polyester, and fluorine-contained acrylic acid resin.

40 10. The process according to claim 8, wherein said thermoplastic resin film has a thickness of from 10 to 200 μm and said porous carbonaceous plate has a mean pore diameter of from 20 to 150 μm and a porosity of from 40 to 85%, said porous carbonaceous plate including short carbon fibers, having a diameter of single fiber of from 4 to 15 μm and a fiber length of from 2 to 20 mm, which are randomly dispersed within substantially a 2-dimensional plane and are bonded with each other by carbonized resin.

FIG. 1

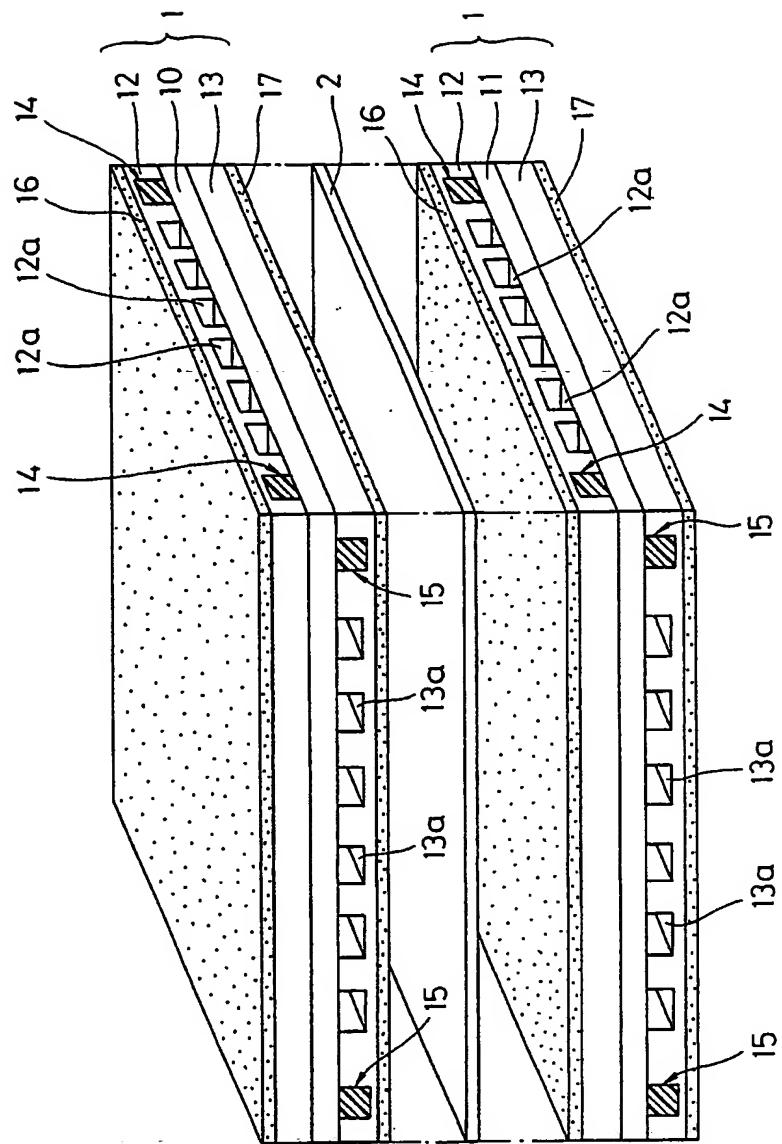


FIG. 2

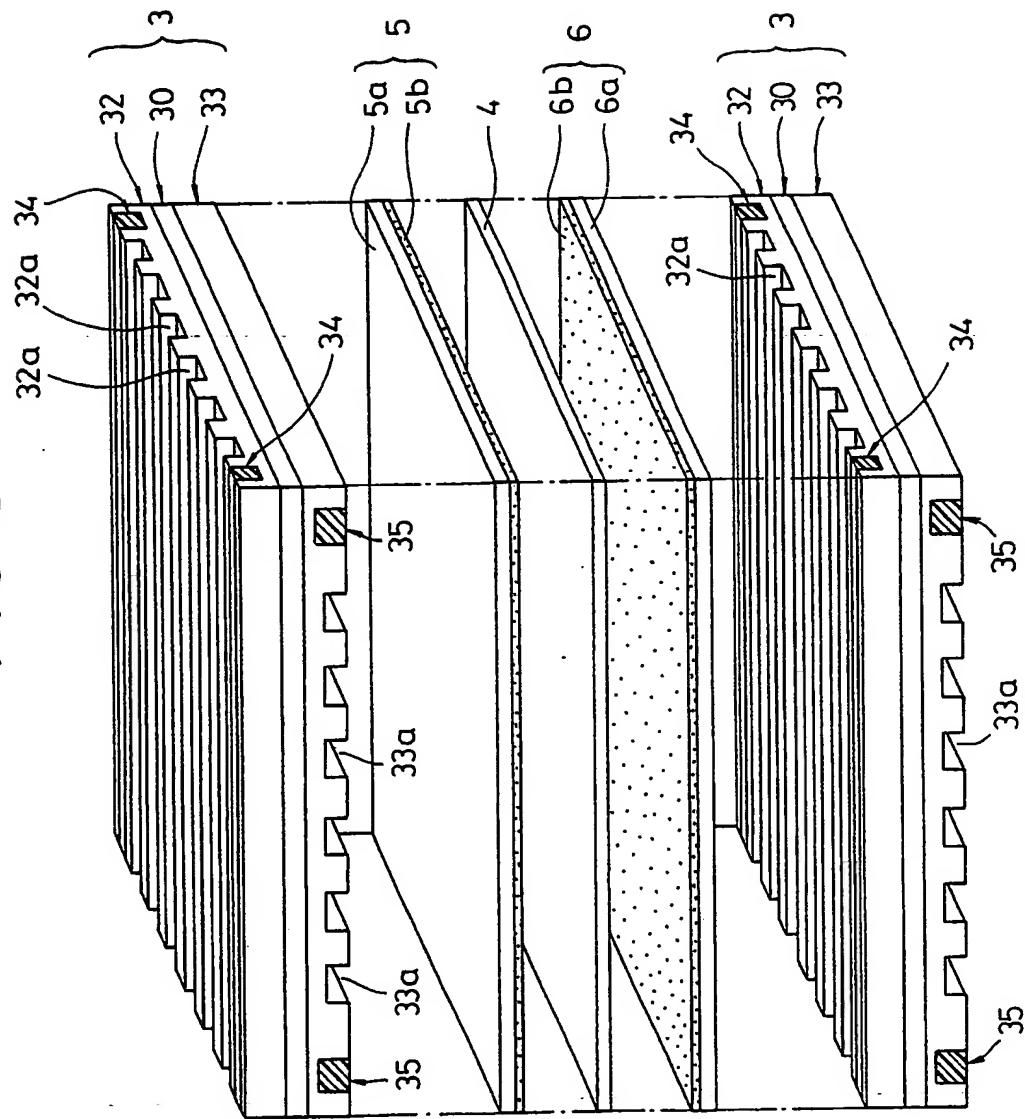


FIG. 3

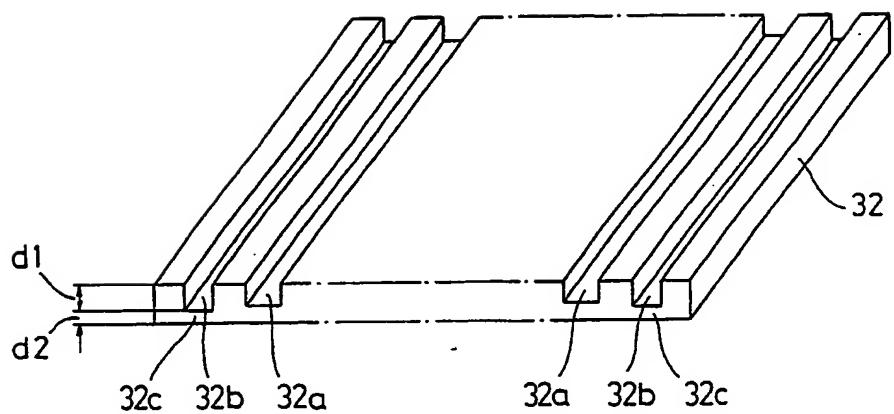


FIG. 4

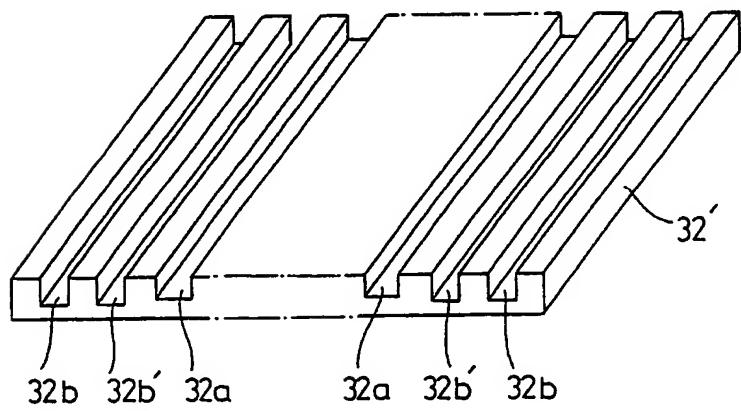


FIG. 5

FIG. 6

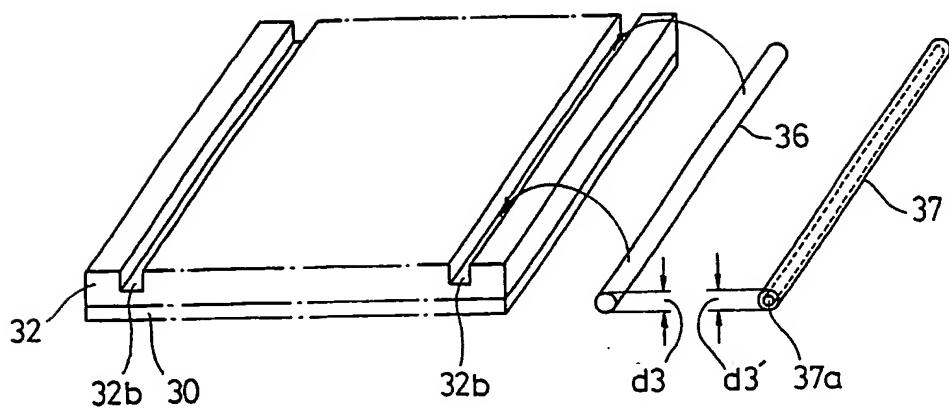


FIG. 7

FIG. 8

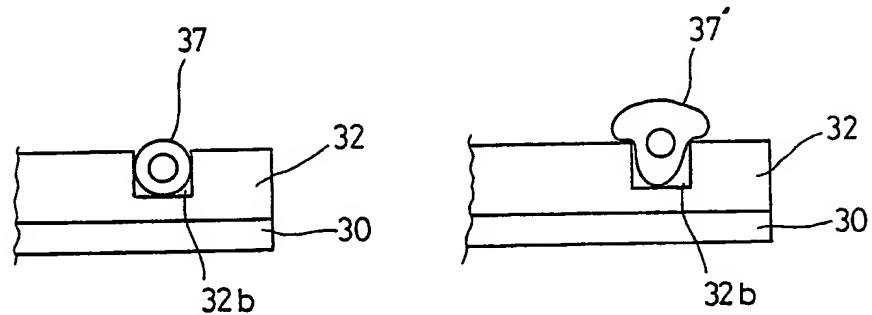


FIG. 9

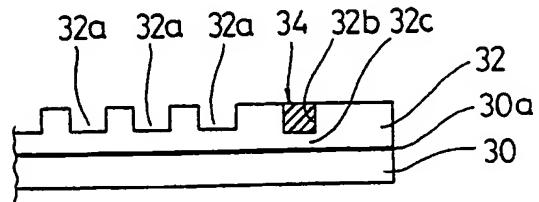


FIG.10

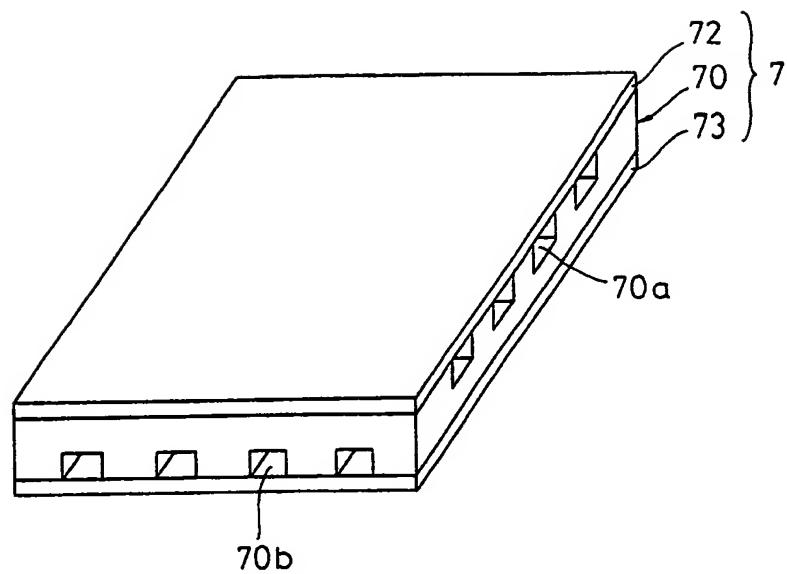


FIG.11

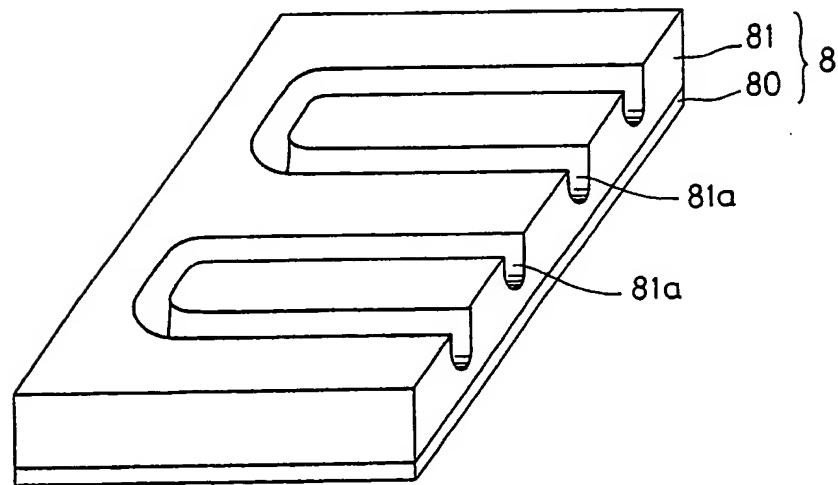
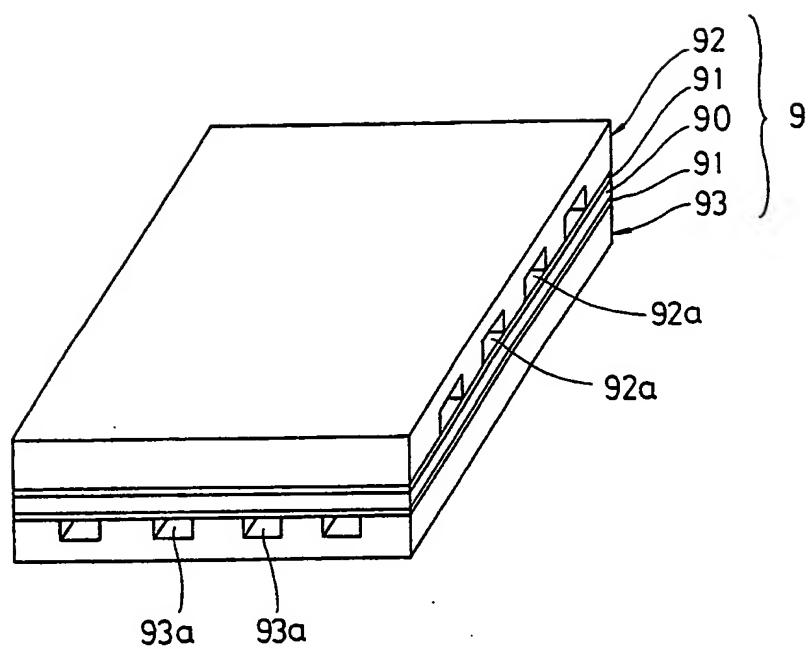


FIG. 12





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⑫

EUROPEAN PATENT APPLICATION

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㉓ Electroconductive integrated substrate and process for producing the same.

㉔ An electroconductive integrated substrate and a process for producing the same, which comprises a porous carbonaceous plate (12, 13) having a large number of pores and gas permeability, and a gas unpermeable electroconductive plate (10) laminated on the porous carbonaceous plate and integrally bonded thereto by thermoplastic resin which is present substantially only in the pores of the porous carbonaceous plate. It is preferred to use thermoplastic resin having a melting viscosity of from 10^2 to 10^6 poises, and a gas permeable porous carbonaceous plate having a mean pore diameter of from 20 to 150 μm and a porosity of from 40 to 85%. The porous carbonaceous plate and the gas unpermeable electroconductive plate are laminated one upon the other with a thermoplastic resin film interposed therebetween, and the film is melted to bond the plates together. The integrated substrate obtained has an electrical resistance of not greater than 7 $\text{m}\Omega$, preferably not greater than 1.5 $\text{m}\Omega$, at the bonded surface in the thickness direction thereof.

EP 0 330 124 A3



DOCUMENTS CONSIDERED TO BE RELEVANT					
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)		
X	EP-A-0 122 150 (ENGELHARD CORP.) * Claims 1,4,5,7,8,10; page 16, line 28 - page 17, line 8; page 20, line 33 - page 21, line 10; page 17, lines 23-32; page 18, line 80 - page 19, line 12 * - - -	1-3,7,8,9	H 01 M 4/96 H 01 M 8/02 C 04 B 37/00		
Y	EP-A-0 174 762 (ENGELHARD CORP.)	4,6			
Y	* Claims 1,2,5,6,7,8,10,11; page 19, lines 5-18; page 21, line 10 - page 22, line 10; page 23, line 20 - page 25, line 5; page 26, line 12 - page 27, line 22 * - - -	6			
A					
Y	GB-A-2 181 422 (KUREHA KAGAKU KOGYO K.K.) * Claim 1; page 8, lines 34-35,45-56; example 1; page 2, lines 13-17 *	1-3,7-9 4			
Y	GB-A-2 193 838 (KUREHA KAGAKU KOGYO K.K.) * Page 5, line 62 - page 6, line 12; page 7, lines 44-59 *	4			
X	GB-A-2 095 150 (ENERGY DEVELOPMENT ASSOCIATES) * Claims 1,3,10,11,15,19; page 2, lines 96-120; page 3, lines 38-40 *	1,2,8,9			
X	EP-A-0 176 247 (ENGELHARD CORP.) * Claims 1,7,8,9,10,11; page 13, line 19 - page 14, line 6; page 15, lines 11-16; page 16, lines 9-29 *	1,2,7,8,9	H 01 M C 04 B		
X	PATENT ABSTRACTS OF JAPAN, vol. 7, no. 276 (E-215)[1421], 9th December 1983; & JP-A-58 157 063 (MITSUBISHI DENKI K.K.) 19-09-1983 * Abstract *	1,2,6,7,8, 9			
The present search report has been drawn up for all claims					
Place of search	Date of completion of search	Examiner			
The Hague	05 March 91	D'HONDT J.W.			
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T: theory or principle underlying the invention					

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EP 89 10 2920

DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.5)
X	US-A-4 245 009 (R.J. GUTHRIE) * Claims 1,3,4; column 5, lines 34-48; column 6, lines 10-16 * - - -	1-3,7-9	
X	DATABASE CHEMICAL ABSTRACTS (HOST:STN), vol. 105, no. 20, 1986, abstract no. 175962d, Columbus, Ohio, US; & JP-A-61 138 464 (HITACHI LTD) 25-06-1986 * Abstract *	1,7	
X	DATABASE CHEMICAL ABSTRACTS (HOST:STN), vol. 107, no. 8, 1987, abstract no. 62063j, Columbus, Ohio, US; & JP-A-62 103 978 (HITACHI LTD) 17-05-1987 * Abstract *	1,2,6,7	
A	GB-A-1 169 418 (VARTA AG) * Claims 1,2,15; page 2, lines 74-76,87-112 * - - -	1,2,7	
A	PATENT ABSTRACTS OF JAPAN, vol. 10, no. 99 (E-396)[2156], 16th April 1986; & JP-A-60 241 654 (HITACHI SEISAKUSHO K.K.) 30-11-1985 * Abstract *	1	TECHNICAL FIELDS SEARCHED (Int. Cl.5)
A	FR-A-2 237 003 (KUREHA KAGAKU KOGYO K.K.) * Claim 1; page 4, lines 5-10; page 5, lines 35-37 * & JP-B-53 018 603 (Cat. D,A) - - - -	4	
The present search report has been drawn up for all claims			
Place of search	Date of completion of search	Examiner	
The Hague	05 March 91	D'HONDT J.W.	
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P: intermediate document			
T: theory or principle underlying the invention			